Register Number:

DATE:

ST. JOSEPH'S COLLEGE (AUTONOMOUS), BENGALURU – 27 B.Sc. – IV SEMESTER END SEMESTER EXAMINATION, APRIL 2020 CH 418 : CHEMISTRY

Time: 1½ hours

Max. Marks: 35

3 X 2 = 6

Note: This question paper has three pages and three sections

PART A

Answer any THREE of the following:

1. State saytzeff's rule.

- 2. Write the Williamson's ether synthesis of diethyl ether.
- 3. Write a chemical reaction following Markonikov's rule.
- 4. Inspite of being a primary alkyl halide, neopentyl iodide undergoes $S_{\text{N}}1$ reaction. Why?
- 5. Identify the nucleophile and the leaving group in the following reactions:

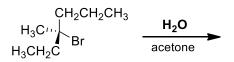
a) $\swarrow \overset{\Theta}{\overset{}}$ + CH₃I \longrightarrow $\checkmark \overset{S}{\overset{}}$ + $\overset{\Theta}{\overset{}}$ b) $\checkmark \overset{\Theta}{\overset{}=}$ + CH₃I \longrightarrow $\checkmark \overset{\Theta}{\overset{}=}$ + $\overset{\Theta}{\overset{}}$

PART B

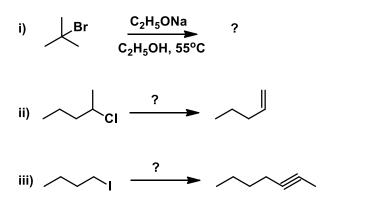
Answer any FOUR of the following:

4 X 6 = 24

6. a) Write the possible product(s) and reaction mechanism for the following reaction: (Hint: Acetone is used as a co-solvent to dissolve the alkyl halide)



b) Complete the following reactions:



(3+3)

- Arrange the following alkyl halides: (CH₃)₃CBr, CH₃Br, CH₃CH₂Br, (CH₃)₂CHBr with respect to their reactivity towards S_N1and S_N2 reactions. Give reasons for their order of reactivity.
- 8. a) Discuss the relative stability of the following alkenes based on their heats of hydrogenation: 1-butene, *cis*-2-butene, *trans*-2-butene. (Numerical values of heats of hydrogenation not required).

b) What is Diels Alder reaction? Give an example. (3+3)

9. a) Write appropriate chemical reactions to show the synthesis of an alkene from:
i) geminal dihalide; ii) vicinal dihalide.
b) Explain the regioselectivity of the reaction of HBr to 1-propene in the presence of

a peroxide. (3+3)

- 10. a) Propose a suitable reaction mechanism for the reaction of 2-butene with bromine. Comment on the stereochemistry of the product.b) Explain the effect of solvent on substitution reactions of alkyl halides. (3+3)
- 11. a) Write the structures of major and minor products formed when HBr is added to 1,3-butadiene at i)-80°C, ii) 40°C. Identify the kinetic and thermodynamic products. (Free energy diagram not required).

b) Explain the regioselectivity of base catalyzed ring opening of an unsymmetrical epoxide with an example. (3+3)

PART C

Answer any ONE of the following:

1 X 5 = 5

(5)

 a) A compound X decolourised a solution of potassium permanganate and produced a brown precipitate. Upon treatment with 1 mole of H₂/Pt, X gave *n*propane. Identify X and write the chemical reactions involved.

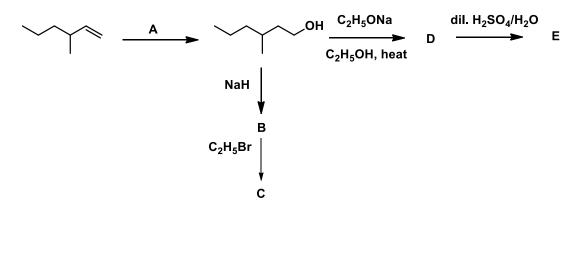
b) Write the products of the following reactions:

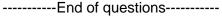
i)
i)
i)
ii)
HC=CH

$$\frac{H_2O/H^+}{HgSO_4}$$
iii)
HC=CH

$$\frac{H_2O/H^+}{HgSO_4}$$
iii)
iiiii)
iiii)
iii)
iii)
iii)
iiii)
iii)
iii)
iiii)
iii)
iii

13. Predict **A** to **E** in the following sequence of reactions:





CH418_B_2020